

"CORROSION IS THE DETERIORATION OF A SUBSTANCE (USUALLY A METAL) OR ITS PROPERTIES BECAUSE OF A REACTION WITH ITS ENVIRONMENT."



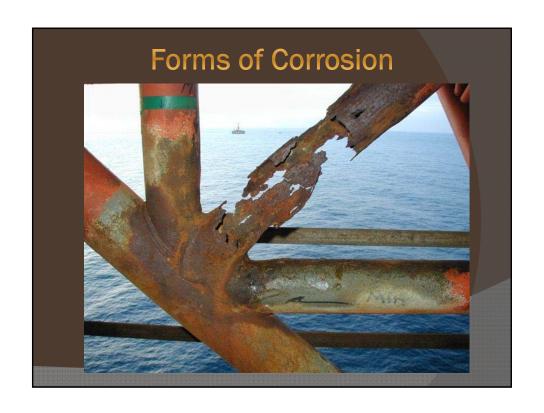








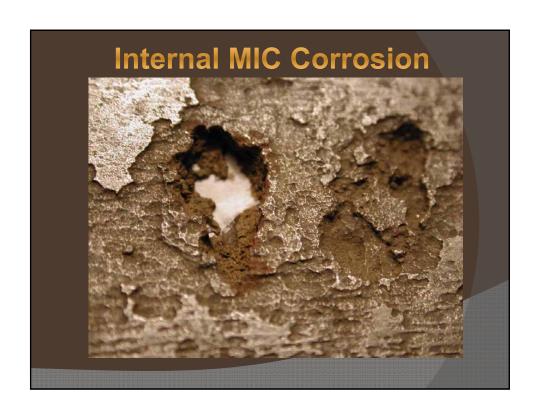


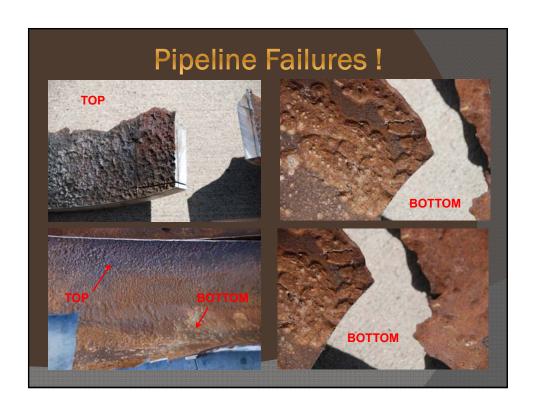


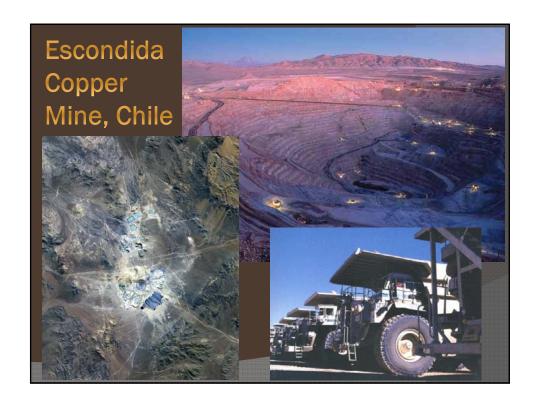




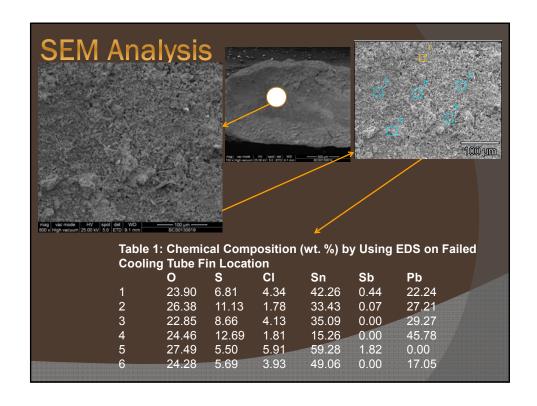






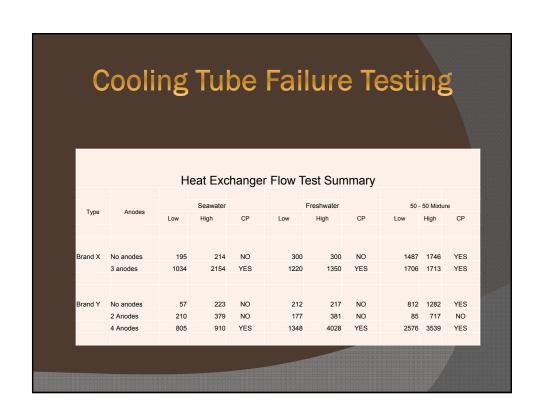


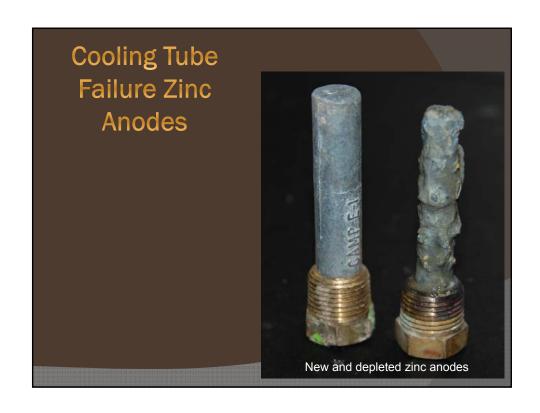














### 7 Methods of Corrosion Control

- Material Selection
- Design of the Structure
- Coatings
- Cathodic Protection
- Alteration of the Environment
- Inhibition
- Repair and/or Replace

### MATERIAL SELECTION

CRA's (Corrosion Resistant Alloys can be used on new structures - BUT - it is expensive, often to the point of not using it.







### **DESIGN OF THE STRUCTURE**

### Proper design can lessen the probability of corrosion:

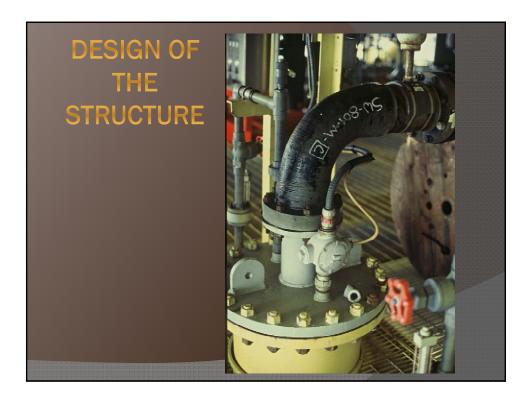
Avoidance of crevices and cracks.

Avoidance of no-flow areas & proper drainage.

Even flow over the surfaces, internal and external.

Proper sizing of pipes, cooling loops, pumps, etc.

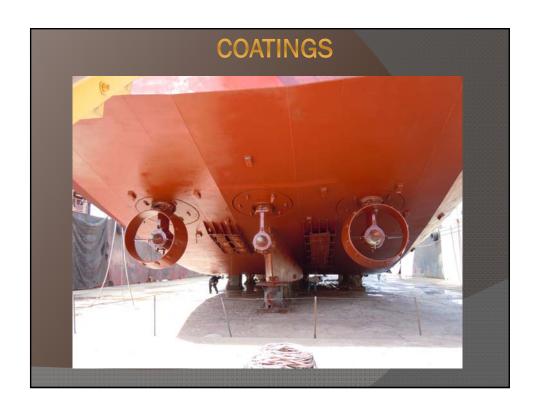
Isolation of different materials.



### **COATINGS**

Coatings are designed to separate the structure from the environment. Inside and outside ... but in coolers it affects the heat exchange properties.



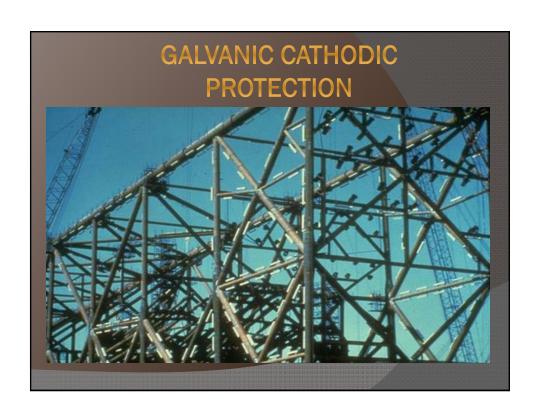


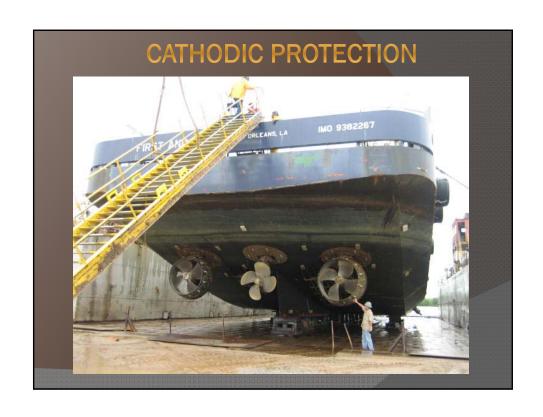
### CATHODIC PROTECTION

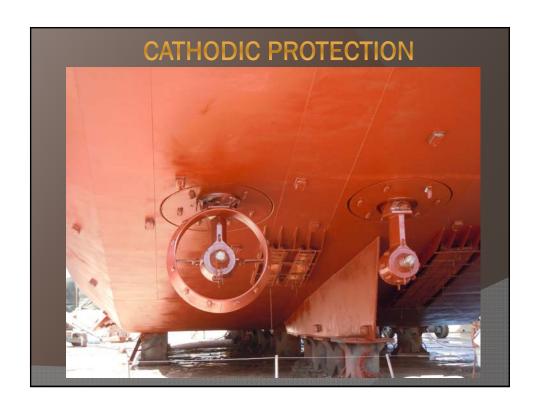
Cathodic Protection (CP) protects the structure by using either Galvanic (Sacrificial) or Impressed Current (ICCP) systems.

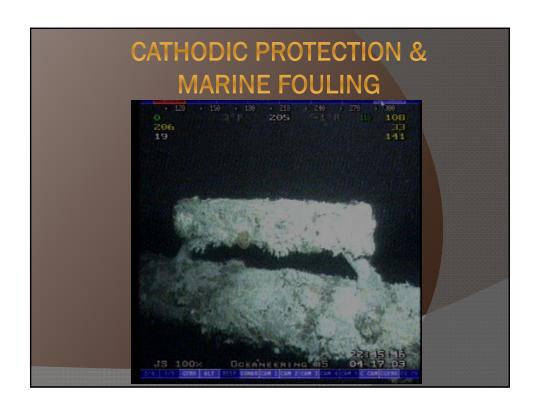
Galvanic CP employs anodes so that the current flow off of the structure goes through the anodes thus they "sacrifice" themselves.

ICCP employs current flowing into the electrolyte through anodes balancing the potential of the electrolyte and the structure so that no current flows off of the structure.

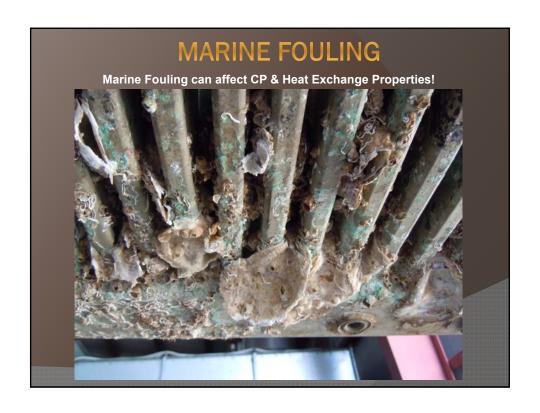




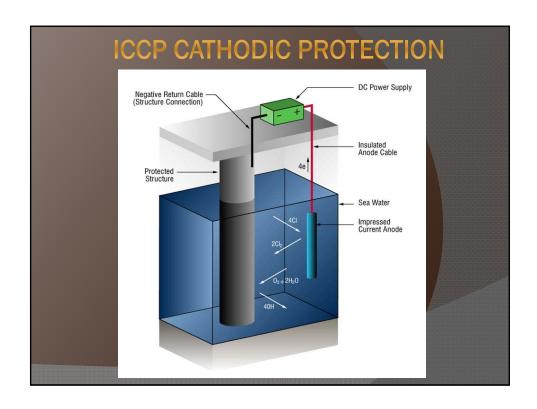












## ALTERATION OF THE ENVIRONMENT - EXTERNAL

Unfortunately the environment is what it is and can not usually be changed externally.

A drill ship built in Pusan sailed to Singapore, Durban and ended up in GOM. 4 different environments.

Deep water GOM vessels start in polluted seawater near the dock and sail into different waters.

Seasonal and location changes.

## ALTERATION OF THE ENVIRONMENT - INTERNAL

The internal environment can often be changed to make it less corrosive and free of marine life, oxygen and chlorides (salt). Closed systems are easier to control that open systems.

Filters and anti-fouling system can be employed.

Sea water can be processed into potable water.

Additives (inhibitors) can be added to create a film between the water and the metal.

Proper drainage.

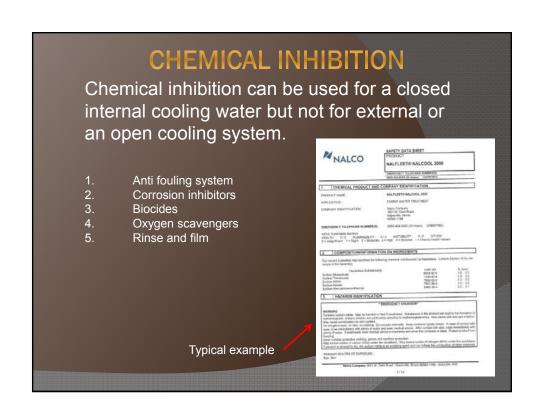
Rinsing salt water with fresh water can greatly lower the corrosion rate.

### **INHIBITION**

Chemical Inhibition is the use of chemicals to alter the environment and/or create a film that separates the water from the metal thus removing contact between the water and the metal.

Oxygen scavengers Biocides Corrosion inhibitors H<sub>2</sub>S scavengers Etc.





# BOTH ONSHORE & OFFSHORE THE INJECTED CHEMICALS HAVE BEEN MONITORED BY:

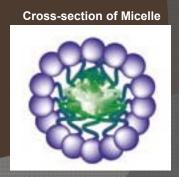
- 1.Leak records
- 2.Coupons & Probes
- 3.Sampling
- 4.Residual checks
- 5.Inspections

Finally a Novel Approach to
Determine if the Chemical
Inhibitors are Effective & for the
Proactive Management of
Internal Corrosion

### This Technique Detects Micelles

A micelle is a nanoscale aggregate of corrosion inhibitor that forms when a system has reached saturation i.e. corrosion inhibitor has adhered to every available surface (pipe, solids & interface).

Micelle



### Micelle Measurement

Micelles are directly measured using a fine stream of fluids passing through a flow cell. A laser then detects the particles and software distinguishes the size, shape, and chemistry of the particle.

We can then analyze the concentration series of the Critical Micelle Concentration (CMC), if required, but this technology relies upon measuring micelles rather than the CMC.

### The Data will give you the following:

- Ensuring optimization of chemical dosage.
- Achieving maximum inhibition.
- Leading to more effective corrosion management currently not offered by other conventional corrosion monitoring methods such as corrosion probes and residual monitoring.

This independent and accurate information provides operators with a clear understanding of the corrosion and associated integrity risks within their critical process equipment, from the wellhead through to onshore receiving facilities, refining and distribution network.

## However.....

Failures have (and are) occurring due to internal corrosion.

Is the inhibitor working?

Is it the correct amount?

Is the monitoring/sampling working?

#### THE LAST METHOD OF CORROSION CONTROL ...

# REPAIR AND/OR REPLACE

As the term implies simply let it corrode and then replace it.

This only works if you have monitoring, inspection and surveillance to determine when to remove it from service before it causes leaks, breakdown or failure.

Note that this is the most widely method of corrosion control across all aspects of industry and commercial uses of metal structures and products.

### 7 Methods of Corrosion Control

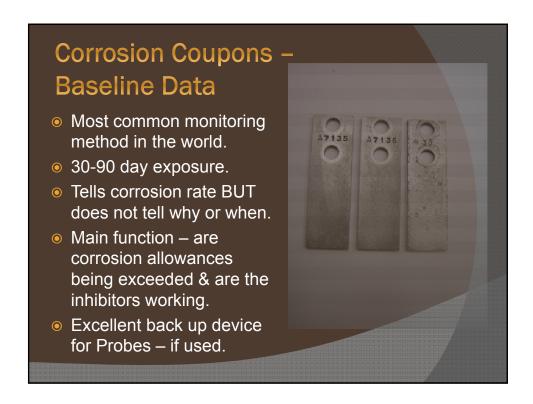
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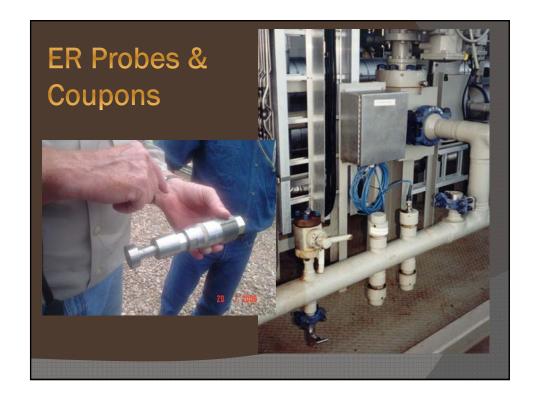
### INSPECTION, MONITORING, SAMPLING & OVERALL SURVEILLANCE

Regardless of your choice (s) of corrosion control, or none at all, you must have a handle on your structures integrity. You can do this through:

Inspections
Monitoring
Sampling
Overall surveillance







Sticking your head in the sand (water) is never the solution.

Good corrosion control & proactive management of heat exchanger corrosion will optimize your integrity efforts & significantly impact your maintenance cost and planning.

